

THE SYNTHESIS AND ABSOLUTE CONFIGURATION  
OF LABALLENIC ACID

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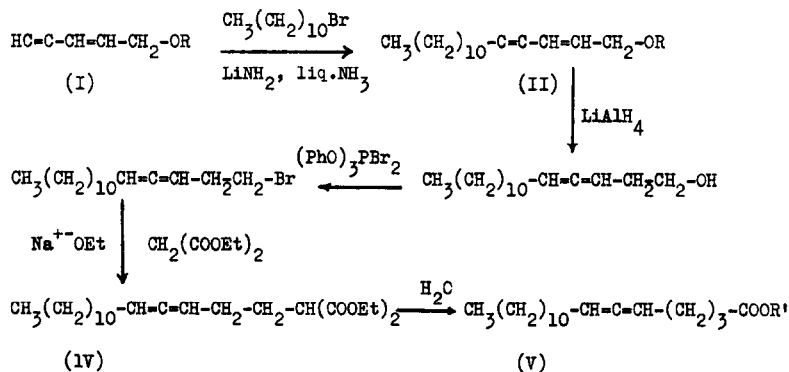
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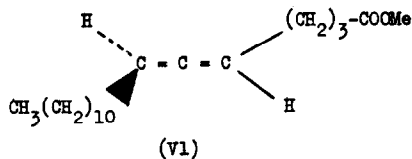
Esters of Laballenic acid, a new optically active eighteen carbon allenic acid, were isolated from the seed oil of *Leonitis nepetaefolia*, L. (R.Br.) of the family of Labiatae. First believed to be the allenic cyclopropane carboxylic acid, 2,3-methylene-heptadeca-4,5-dienoic acid<sup>1</sup>, it has recently been shown to be octadeca-5,6-dienoic acid<sup>2</sup>. We have synthesised octadeca-5,6-dienoic acid by the following method:-

1-Tetrahydropyranyloxy-pent-2-en-4-yne (I) with lithium in liquid ammonia and n-undecyl bromide gave 1-tetrahydropyranyloxy-hexadec-2-en-4-yne (II, R = OC<sub>6</sub>H<sub>9</sub>) which was hydrolysed to hexadec-2-en-4-yn-1-ol (II, R = H). This was reduced with lithium aluminium hydride<sup>3</sup> to give hexadeca-3,4-dien-1-ol (III) which was converted to the corresponding bromide by the triphenylphosphite-dibromide method<sup>4</sup>. 1-Bromohexadeca-3,4-diene and diethyl malonate gave the hexadeca-3,4-dienylmalonic ester (IV) which was hydrolysed and decarboxylated to give laballenic acid (V, R' = H),  $\nu_{\max}$  3500-3200 (broad), 1960 m, 1730 s cm<sup>-1</sup>; p-bromophenacyl laballenate,

m.p. 46-46.5° (lit.<sup>2</sup> m.p. 47-47.5°); methyl ester (V, R' = Me) b.p. 144-146°/5x10<sup>-2</sup> mm.,  $\nu_{\max}$  1750 s, 1960 m cm<sup>-1</sup>, g.l.c. using silicon oil at 176°, N<sub>2</sub> 1.6 l.hr<sup>-1</sup> gave one component at t 74 min.



When (II, R = H) was reduced with lithium aluminium hydride-3-O-benzyl-1,2-cyclohexylidene- $\alpha$ -D-glucopyranose complex<sup>5</sup>, (-)-hexadeca-3,4-dien-1-ol,  $[\alpha]_{\text{D}}^{20} -4.4^\circ$  was obtained. It has been shown<sup>6</sup> that reduction of 2-en-4-yn-1-ols with this complex gives allenic alcohols of the (R)-configuration, and (-)-hexadeca-3,4-dien-1-ol must therefore be the (R)-form (VI). Conversion via the bromide to the malonic ester (IV) and hydrolysis and decarboxylation gave (-)-laballenic acid (V)  $[\alpha]_{\text{D}}^{20} -3.0^\circ$  and the (-)-methyl ester  $[\alpha]_{\text{D}}^{20} -3.0^\circ$ , and this therefore has the (R)-configuration.



(R)-(-) laballenic acid

As the methyl ester from the naturally occurring laballenic acid is described as having  $[\alpha]_D^{20} - 47.5^\circ$  it must have the (R)-configuration.

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